

$$dW = K(P_A - P_B)dA = -Q_A H dP_A = Q_B H dP_B \quad (1)$$

$$K(P_A - P_B)dA = -Q_A H dP_A$$

$$\frac{dP_A}{P_A - P_B} = -\frac{K dA}{Q_A H}$$

$$K(P_A - P_B)dA = Q_B H dP_B$$

$$\frac{dP_B}{P_A - P_B} = \frac{K dA}{Q_B H}$$

$$\frac{dP_A}{P_A - P_B} - \frac{dP_B}{P_A - P_B} = -\frac{K dA}{H} \frac{1}{Q_A} - \frac{K dA}{H} \frac{1}{Q_B}$$

$$\frac{d(P_A - P_B)}{P_A - P_B} = -\frac{K dA}{H} \left[\frac{1}{Q_A} + \frac{1}{Q_B} \right]$$

$$\ln \frac{(P_{A0} - P_{B0})}{(P_{Ai} - P_{Bi})} = -\frac{K A}{H} \left[\frac{1}{Q_A} + \frac{1}{Q_B} \right]$$

$$\ln \frac{(P_{Gi} - P_{Bi})}{(P_{G2} - P_{B2})} = \frac{KA}{H} \left[\frac{1}{O_B} + \frac{1}{O_G} \right]$$

$$dW = -Q_G H dP_G$$

$$dW = Q_B H dP_B$$

$$W = -Q_G H [P_{G2} - P_{Gi}]$$

$$W = Q_B H [P_{B2} - P_{Bi}]$$

$$\frac{1}{O_G} = \frac{H(P_{Gi} - P_{G2})}{W}$$

$$\frac{1}{O_B} = \frac{H(P_{B2} - P_{Bi})}{W}$$

$$\ln \frac{(P_{Gi} - P_{Bi})}{(P_{G2} - P_{B2})} = \frac{KA}{H} \left[\frac{H(P_{B2} - P_{Bi})}{W} + \frac{H(P_{Gi} - P_{G2})}{W} \right]$$

$$W = \frac{KA [(P_{Gi} - P_{Bi}) - (P_{G2} - P_{B2})]}{\ln \frac{(P_{Gi} - P_{Bi})}{(P_{G2} - P_{B2})}}$$

= equazione logaritmo medio ossigenazione.

$P_{GiO_2} = 760 - 47 \text{ mmHg}$	}	$= 713 \text{ mmHg}$
$P_{BiO_2} = 40 \text{ mmHg}$		$P_{BiCO_2} = 46 \text{ mmHg}$
$P_{B2O_2} = 104 \text{ mmHg}$		$P_{B2CO_2} = 40 \text{ mmHg}$
$P_{G2O_2} = 713 - 64 =$		

$$W_{O_2} = 250 \frac{\text{ml}}{\text{min}}$$

$$P_{GiCO_2} = \phi$$

$$W_{CO_2} = 200 \frac{\text{ml}}{\text{min}}$$

$$P_{G2CO_2} = 6 \text{ mmHg}$$

$$A_{OTT} = \frac{A_{O_2} + A_{CO_2}}{2}$$

$$O = \text{Oxygenation} = Q_B \frac{P_{Bi} - P_{Bo}}{P_{Bi} - P_{Gi}}$$

$$D = Q_B \frac{C_{Bi} - C_{Bo}}{C_{Bi} - C_{Di}}$$

$$C = D|_{C_{Di}} \quad C_{Ox} = O|_{P_{Gi} = P} = Q_B \frac{P_{Bi} - P_{Bo}}{P_{Bi}} \quad \text{total } \frac{269}{40}$$

$$E = \frac{D}{Q_B} \Big|_{C_{Di} = P} \quad E_{Ox} = \frac{O}{Q_B} \Big|_{C_{Di} = P} = \frac{P_{Bi} - P_{Bo}}{P_{Bi}} = 1 - \frac{P_{Bo}}{P_{Bi}} \quad O_D \gg O_B$$

$$E = 1 - e^{-N_T} \quad N_T = \frac{KA}{Q_B}$$

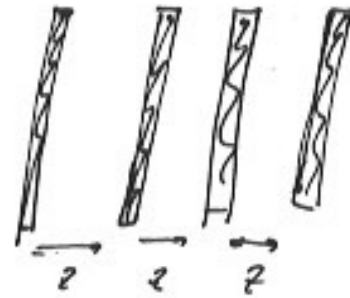
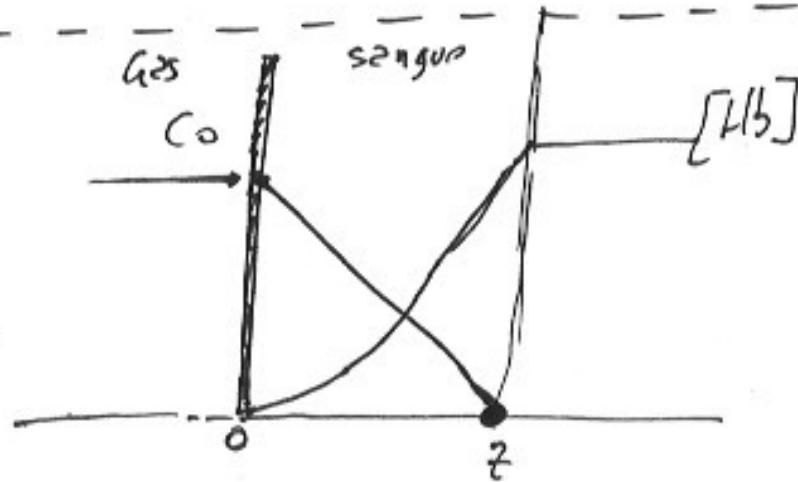
$$E = 1 - e^{-N_T} = 1 - \frac{P_{Bo}}{P_{Bi}} \quad \underline{P_{Bo} = P_{Bi} e^{-N_T}}$$

$$V_b \frac{dP_{Bi}}{dt} = Q_B (P_{Bo} - P_{Bi}) = Q_B (P_{Bi} e^{-N_T} - P_{Bi}) = Q_B P_{Bi} (e^{-N_T} - 1) =$$

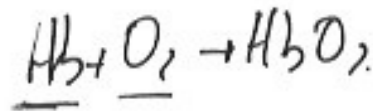
$$e^{-N_T} = \beta \quad V_b \frac{dP_{Bi}}{dt} = Q_B P_{Bi} (\beta - 1) \quad \frac{dP_{Bi}}{P_{Bi}} = \frac{Q_B (\beta - 1)}{V_b} dt$$

$$\ln \frac{P_{Bi}^R}{P_{Bi}^I} = \frac{Q_B}{V_b} (\beta - 1) t \quad \rightarrow \quad \underline{P_{Bi}^R = P_{Bi}^I e^{\frac{Q_B}{V_b} (\beta - 1) t}}$$

where β is temperature assignation ⁽⁴⁾



$$-DA \frac{dc}{dz} = -DA \frac{C_B}{z}$$

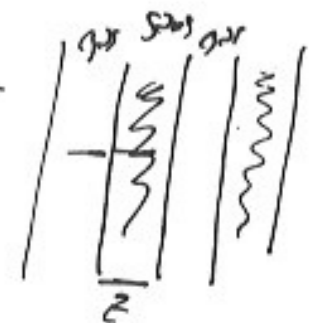


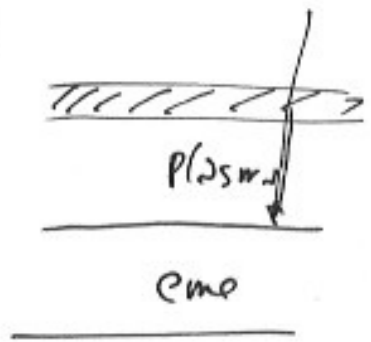
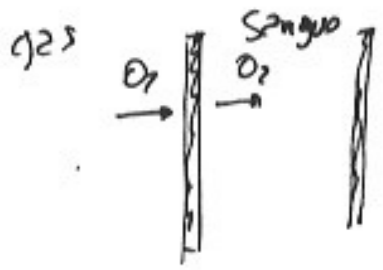
$$+ DA \frac{C_B}{z} dt = t [Hb] A dz$$

$$+ \frac{D C_B dt}{[Hb]} = z dz$$

$$\frac{D C_B t}{[Hb]} = \frac{z^2}{2}$$

$$z = \sqrt{\frac{2 D C_B t}{[Hb]}}$$





$$\frac{\partial c'}{\partial t} = D_2 \frac{\partial^2 c'}{\partial x^2}$$

$$\frac{\partial c}{\partial t} = D_1 \frac{\partial^2 c}{\partial x^2} + K(y_0 - y) - K'cy$$

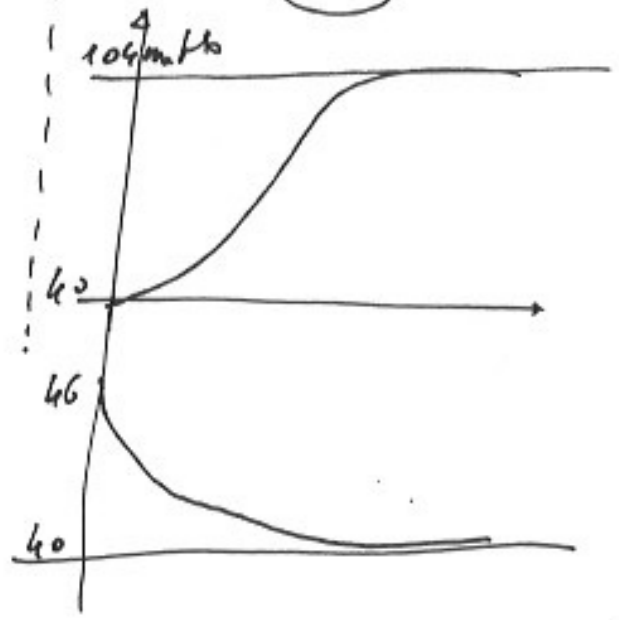
$$\frac{\partial y}{\partial t} = D_{H_2O} \frac{\partial^2 y}{\partial x^2} + k(y_0 - y) - k'cy$$

$$\frac{\partial c}{\partial t} = A$$

$$D_1 \frac{\partial^2 c}{\partial x^2} = A$$

$$\frac{\partial c}{\partial t} = D_1 \frac{\partial^2 c}{\partial x^2}$$

$$\frac{1}{D_2} = \frac{1}{D_1} + \frac{1}{D_3}$$



$$\frac{\partial c}{\partial t} = \frac{D_{O_2}}{D_{O_2}} \frac{\partial^2 c}{\partial x^2} + R_{O_2}$$

$$D_{H_2} = 0$$

$$\frac{\partial y}{\partial t} = D_{H_2} \frac{\partial^2 y}{\partial x^2} + R_{H_2}$$

$$R_{O_2} = \frac{m d i}{\text{cm}^3 \cdot \text{sec}}$$

$$\frac{\partial c}{\partial t} = \phi \quad \begin{cases} \frac{\partial^2 c}{\partial x^2} = -\frac{R_{O_2}}{D_{O_2}} \\ \frac{\partial y}{\partial t} = R_{H_2} \end{cases}$$

$$c(x) \propto x^2$$

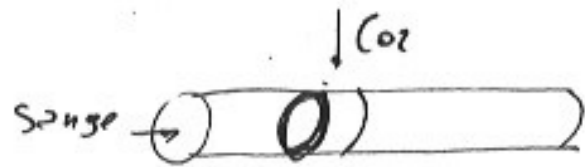
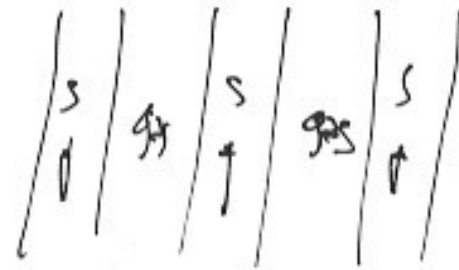
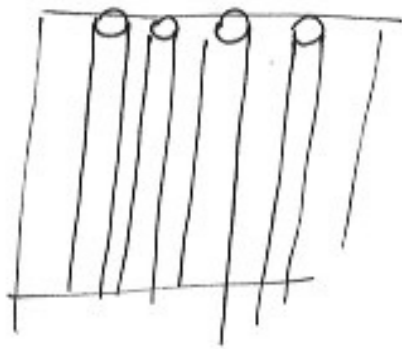
$$y \propto t$$

$$R_{O_2} + R_{H_2} = \phi$$

$$\frac{\partial^2 c}{\partial x^2} + \frac{\partial y}{\partial t} = -\frac{R_{O_2}}{D_{O_2}} + R_{H_2} = -R_{O_2} \left(\frac{1}{D_{O_2}} + \frac{R_{H_2}}{R_{O_2}} \right) = A$$

$$\frac{\partial^2 c}{\partial x^2} + \frac{\partial y}{\partial t} = A = \text{const.} =$$





$$J_{O_2} = -D \frac{\partial c}{\partial r} = 250 \frac{\text{mol}}{\text{m}^2 \cdot \text{s}}$$

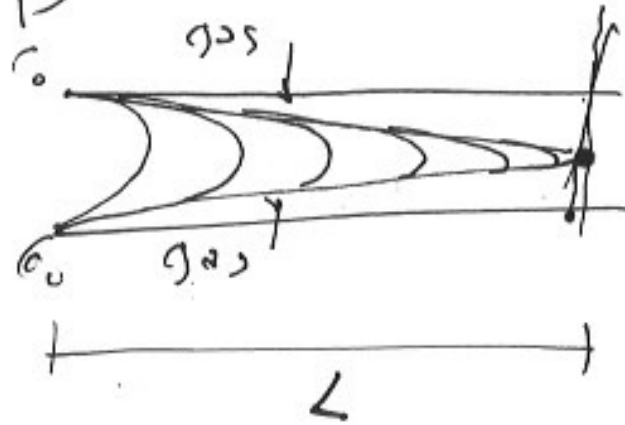
$$J_{O_2} \cdot 2\pi r = A =$$



$$-D \frac{\partial c}{\partial r} \cdot 2\pi r = A$$

$$\partial c = -\frac{A}{D} \cdot 2\pi \frac{dr}{r}$$

$$c(r) - c(r_f) = -\frac{A}{D} 2\pi \ln \frac{r}{r_f}$$



$$\underline{c(r) = c_0 - \frac{A}{D} 2\pi \ln \frac{r}{r_f}}$$